Distinct Source Water Chemistry Shapes Contrasting Concentration-Discharge Patterns

Wei Zhi, Li Li, Wenming Dong, Wendy Brown, Jason Kaye, Carl Steefel, Kenneth H. Williams

1Department of Energy and Mineral Engineering, The Pennsylvania State University, University Park, PA
2Department of Civil and Environmental Engineering, The Pennsylvania State University, University Park, PA
3Earth and Environmental Sciences Area, Lawrence Berkeley National Laboratory, Berkeley, CA
4The Rocky Mountain Biological Laboratory, Crested Butte, CO
5Department of Ecosystem Science and Management, The Pennsylvania State University, University Park, PA

Project Lead Principal Investigator (PI): Li Li, SBR University project DE-SC0016221, collaboration with the Watershed Function Science Focus Area at Lawrence Berkeley National Laboratory funded by the DOE SBR under the contract DE-AC02-05CH11231

Project Website: Li Reactive Water Group
Contact: lili@engr.psu.edu

Abstract Understanding concentration-discharge (C-Q) relationships are essential for predicting chemical weathering and biogeochemical cycling under changing climate and anthropogenic conditions. Contrasting C-Q relationships have been observed widely, yet a mechanistic framework that can interpret diverse patterns remains elusive. This work hypothesizes that seemingly disparate C-Q patterns are driven by switching dominance of end-member source waters and their chemical contrasts arising from subsurface biogeochemical heterogeneity. We use data from Coal Creek, a high-elevation mountainous catchment in Colorado, and a recently developed watershed reactive transport model (Biort-Flux-PhHM). Sensitivity analysis and Monte-Carlo simulations (500 cases) show that reaction kinetics and thermodynamics and distribution of source materials across depths govern the chemistry gradients of shallow soil water and deeper groundwater entering the stream. The alternating dominance of organic-poor yet geo-solute-rich groundwater under dry conditions and organic-rich yet geo-solute-poor soil water during spring melt leads to the flushing pattern of dissolved organic carbon and the dilution pattern of geogenic solutes (e.g., Na, Ca, and Mg). In addition, the extent of concentration contrasts regulates the power law slopes (b) of C-Q patterns via a general equation \( b = \frac{\delta_b C_{\text{ratio}}}{C_{\text{ratio}}} + b_{\text{min}} \). At low ratios of soil water versus groundwater concentrations (\( C_{\text{ratio}} = C_{\text{sw}} / C_{\text{gw}} < 0.6 \)), dilution occurs; at high ratios (\( C_{\text{ratio}} > 1.8 \)), flushing arises; chemostasis occurs in between. This equation quantitatively interprets b values of 11 solutes (dissolved organic carbon, dissolved P, NO\(_3\), K, Si, Ca, Mg, Na, Al, Mn, and Fe) from three catchments (Coal Creek, Shale Hills, and Plynlimon) of differing climate, geologic, and land cover conditions. This indicates potentially broad regulation of subsurface biogeochemical heterogeneity in determining C-Q patterns and wide applications of this equation in quantifying b values, which can have broad implications for predicting chemical weathering and biogeochemical transformation at the watershed scale.

Reference: